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Tetrahedron Letters 47 (2006) 4929–4932

Tetrahedron Letters

Direct one-pot highly enantioselective assembly of polyketide and carbohydrate synthons

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Received 13 March 2006; revised 23 April 2006; accepted 4 May 2006 Available online 26 May 2006

Abstract—A short, direct, catalytic, enantioselective synthesis of polyketide segments and carbohydrates is presented. The novel, direct, one-pot, organocatalytic asymmetric tandem cross-aldol/Horner–Wittig–Emmons reactions assemble polyketide and carbohydrate derivatives in good yield with 93–98% ee. The one-pot catalytic asymmetric tandem reaction was applied to a highly enantioselective formal de novo synthesis of the rare carbohydrate, L-altrose. © 2006 Elsevier Ltd. All rights reserved.

The rapidly growing development of glycobiology and carbohydrate based pharmaceuticals demands the increased development of reaction design and method advancement for the selective construction of natural and unnatural carbohydrates.^{[1](#page-2-0)} In this context, the de novo synthesis of carbohydrates is of immense impor- $tance.^{2,3}$ $tance.^{2,3}$ $tance.^{2,3}$ However, most conventional monosaccharide

of biomimetic, catalytic, asymmetric, one-pot and tan-dem reactions.^{[9,10](#page-3-0)} Based on retrosynthetic analyses, we envisioned that a one-pot, direct, asymmetric, tandem aldol-Horner–Wittig–Emmons (HWE) sequence would be an excellent catalytic entry for the enantioselective synthesis of polyketide and carbohydrate derivatives (Eq. 1).

H O R1 + Amino Acid R OH ^R ^R ¹ OH R1 O H HWE R H O O OR² ð1Þ

syntheses start from the chiral pool and involve multiple steps and require protective group strategies and subsequent reduction–oxidation steps.[4](#page-2-0) The reduction of protective group strategies can be accomplished by the use of enzymes as catalysts for the highly selective de novo synthesis of carbohydrates^{[5](#page-2-0)} and polyketide natural products.[6](#page-2-0) Organocatalysis is a rapidly expanding research field within asymmetric catalysis.^{[7](#page-2-0)} Most recently, amino acid catalysis was added to the repertoire of carbohydrate synthesis.[8](#page-2-0) Inspired by Nature's ability to assemble complex molecules from simple precursors in an asymmetric and economic fashion,^{[6](#page-2-0)} we have begun a research programme that focuses on the development

Herein, we describe a novel, direct, one-pot, organocatalytic asymmetric aldol-HWE reaction that yields polyketide and carbohydrate derivatives in good yield with 93–98% ee.

In an initial experiment, propionaldehyde (2 mmol) was added to a stirred solution of (S) -proline $(10 \text{ mol } \%)$ in DMF (0.5 mL). After 15 h of vigorous stirring at 4° C, the temperature was increased to room temperature and LiBr (1.5 mmol), methyl 2-(diethoxyphosphoryl) acetate (1.5 mmol) and DBU (228 mg, 1.5 mmol) were added to the reaction mixture. After stirring for 4 h at room temperature, the reaction mixture was passed directly through a silica gel column to give the pure natural product mycinonic acid ester 2a in 52% yield with 2.3:1 dr (*anti:syn*) and 93% ee (Eq. [2\)](#page-1-0).^{[11,12](#page-3-0)}

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Encouraged by this result, we investigated the one-pot direct organocatalytic asymmetric tandem aldol/HWE reaction with a set of different aldehydes 1 as substrates (Table 1).

The direct one-pot catalytic asymmetric reactions proceeded smoothly and were highly chemo- and enantioselective. For instance, the catalytic enantioselective reactions with acceptor aldehydes 1b and 1c gave the corresponding α , β -unsaturated- δ -hydroxy esters 2b with an 11:1 dr and 96% ee, and 2c with a 10:1 dr and 98% ee, respectively.[13](#page-3-0) Thus, the reactions with aldehydes 1b and 1c as acceptors were slightly more stereoselective as compared to the reaction with propionaldehyde 1a. The δ -hydroxytriketide products 2 are useful synthons for the synthesis of polyketide macrolactones. Moreover, the direct one-pot organocatalytic asymmetric reaction was also performed with a-benzyloxyacetaldehyde 1d as the donor to give carbohydrate derivative 2d in 60% yield with a 4:1 dr and 98% ee.^{[14](#page-3-0)} In order to

show further the synthetic utility of the amino acidcatalyzed one-pot direct asymmetric tandem crossaldol/HWE reaction, carbohydrate derivative 2d was converted in one step, with high diastereoselectivity, to altronic acid 3d with a 98% ee by catalytic dihydroxylation (Scheme 1).[15,16](#page-3-0) Thus, the novel catalytic one-pot reaction is a direct asymmetric entry to altronic acid derivatives. Next, altronic acid 3d was converted to the corresponding altronic δ -lactone 4d. Subsequent dibenzylation furnished perbenzylated δ -lactone 4d' and comparison of the optical rotation $([\alpha]_D^{25} + 10.1$ (c 0.6, CHCl₃); literature: $[\alpha]_D^{24} + 14.3$ (c 1.01, CHCl₃)^{[16](#page-3-0)}) with the literature established that L -altronic acid δ -lactone 4d' had been assembled in an asymmetric fashion by (S) -proline catalysis.¹⁷ Hence, the stereochemical outcome is in accordance with previous (S)-proline cata-lyzed direct cross-aldol reactions.^{[8](#page-2-0)}

Furthermore, we performed a two-step, organocatalytic, asymmetric, self-aldol, Wittig sequence that gave isomer

Table 1. One-pot, direct, organocatalytic, asymmetric, tandem cross-aldol/HWE reactions

	R^2	÷ н R ¹	1. (S) -proline (10 mol\%) DMF, 4° C 2. DBU, LiBr, $EIO \rightarrow BIO \rightarrow BIO$ EtO ^V	OН R R 2	OMe	
Entry	R	ĸ	Product	Yield ^a $(\%)$	dr'	ee^{c} (%)
	Et	Me	2a	52	2.3:1	93
	i -Pr	Me	2 _b	58	11:1	96
	c -Hexyl	Me	2c	59	10:1	98
	CH ₂ OBn	OBn	2d	60 ^d	4:1	98

^a Isolated yield of pure product 2 after silica gel column chromatography.
^b Determined by NMR analyses.

^c Determined by chiral phase GC or HPLC analyses.

^d The whole reaction sequence was performed at room temperature.

Scheme 1. Reagents and conditions: (a) cat. (S)-proline, aldehyde 1d, DMF, $4 °C$, $48 h$ then (EtO)₂POCH₂COOMe, LiCl, DBU, rt, 60% for 2d; (b) cat. OsO₄, NMO, acetone–H₂O 1:1, rt, 74% for 3d >19:1 dr; (c) cat. CF₃COOH, CH₂Cl₂, rt, 80% for 4d; (d) BnBr, n-Bu₄NI, NaH, THF, rt, 56% for 4d'.

Scheme 2. Reagents and conditions: (a) (i) cat. (S)-proline, aldehyde 1d, DMF, rt, 48 h; (ii) (Ph)₃PCHCOOMe, MeOH, rt; (b) cat. p-TsOH, CH₂Cl₂, rt, 51% (three steps) for 5, 3:1 dr (trans:cis 3:1), 97% ee; (c) cat. OsO4, NMO, acetone–H2O 1:1, rt, 60% for 6 >19:1 dr; (d) BnBr, n-Bu4NI, NaH, THF, rt, 50% for 7.

2d' in a 3:1 (Z :E) ratio with 3:1 dr and 97% ee (Scheme 2).

Lactonization in the presence of p-toluenesulfonic acid gave δ -lactone 5 in a 51% overall yield (three steps) with 97% ee.[18](#page-3-0) Highly diastereoselective dihydroxylation $(>19:1 \text{ dr})^{15}$ $(>19:1 \text{ dr})^{15}$ $(>19:1 \text{ dr})^{15}$ afforded mannoic acid δ -lactone 6 that was dibenzylated to give the corresponding perbenzylated carbohydrate $7^{19,20}$ $7^{19,20}$ $7^{19,20}$ Comparison of the optical rotation $[\alpha]_D^{25}$ -1.0 (c 0.1, CHCl₃); literature: *ent-*7 $[\alpha]_D^{24} + 0.37$ (c 0.1, CHCl₃)^{[17](#page-3-0)} and the NMR data of 7 to the literature confirmed that L -mannoic δ -lactone is furnished by (S) -proline catalysis.^{[20](#page-3-0)}

In summary, we have developed a direct organocatalytic tandem asymmetric cross-aldol-HWE reaction. The reaction proceeds with excellent chemo- and enantioselectivity and furnishes polyketide and carbohydrate derivatives in good yield with up to 10:1 dr and 93– 98% ee. Moreover, altronic acid and mannoic δ -lactone with 98% and 97% ee, respectively, were synthesized by subsequent highly diastereoselective catalytic dihydroxylation. Thus, the combination of biomimetic one-pot catalytic asymmetric reactions and diastereoselective catalytic dihydroxylation is a direct route to carbohydrate derivatives.

Acknowledgements

A.C. thanks the Swedish Research Council, Carl-Trygger and Hierta Foundation, for financial support.

References and notes

- 1. Glycochemistry: Principles; Wang, P., Bertozzi, C., Eds. Synthesis and Applications; Marcel Dekker, 2001.
- 2. For synthesis of ketoses, see: (a) Majewski, M.; Nowak, P. J. Org. Chem. 2000, 65, 5152; (b) Majewski, M.; Nowak, P. Synlett 1999, 1447; (c) Enders, D.; Ince, S. J. Synthesis 2002, 619; (d) Enders, D.; Voith, M.; Ince, S. J. Synthesis 1996, 1775; (e) Enders, D.; Prokopenko, O. F.; Raabe, G.; Runsink, J. Synthesis 1996, 1095; For an excellent review on the use of protected dihydroxy acetones in synthesis,

see: (f) Enders, D.; Voith, M.; Lenzen, A. Angew. Chem., Int. Ed. 2005, 44, 1330, and references cited therein.

- 3. For the synthesis of aldoses see: (a) Nicolaou, K. C.; Mitchel, H. J. Angew. Chem., Int. Ed. 2001, 40, 1576; (b) Evans, D. A.; Hu, E.; Tedrow, J. S. Org. Lett. 2001, 3, 3133; (c) Davies, S. G.; Nicholson, R. L.; Smith, A. D. Synlett 2002, 1637; (d) Danishefsky, S. J.; Maring, C. J. J. Am. Chem. Soc. 1985, 107, 7761; (e) Davies, S. G.; Nicholson, R. L.; Smith, A. D. Org. Biomol. Chem. 2004, 2, 3385; (f) Ko, S. Y.; Lee, A. W. M.; Masamune, S.; Reed, L. A.; Sharpless, K. B.; Walker, W. J. Science 1983, 220, 249; (g) Ahmed, M. M.; Berry, B. P.; Hunter, T. J.; Tomcik, D. J.; O'Doherty, G. A. Org. Lett. 2005, 7, 745.
- 4. Hanessian, S. Total Synthesis of Natural Products: The 'Chiron' Approach; Pergamon Press: Oxford, 1983.
- 5. (a) Fessner, W.-D. In Stereoselective Biocatalysis; Patel, R. N., Ed.; Marcel Dekker: New York, 2000; p 239; (b) Machajewski, T. D.; Wong, C. -H. Angew. Chem., Int. Ed. 2000, 39, 1352; (c) Heine, A.; DeSantis, G.; Luz, J. G.; Mitchell, M.; Wong, C.-H.; Wilson, I. A. Science 2001, 294, 369.
- 6. (a) Khosla, C.; Harbury, P. B. Nature 2001, 409, 247; (b) Kinoshita, K.; Willard, P. G.; Khosla, C.; Cane, D. E. J. Am. Chem. Soc. 2001, 123, 2495, and references cited therein.
- 7. Reviews see: (a) Dalko, P. I.; Moisan, L. Angew. Chem., Int. Ed. 2001, 40, 3726; (b) List, B. Tetrahedron 2002, 58, 5573; (c) Dalko, P. I.; Moisan, L. Angew. Chem., Int. Ed. 2004, 43, 5138; For a summary of amino acid-catalyzed sugar synthesis see: (d) Kazmaier, U. Angew. Chem. 2005, 44, 2186.
- 8. (a) Northrup, A. B.; MacMillan, D. W. C. Science 2004, 305, 1752; (b) Casas, J.; Engqvist, M.; Ibrahem, I.; Kaynak, B.; Córdova, A. Angew. Chem., Int. Ed. 2005, 44, 1343; (c) Reyes, E.; Córdova, A. Tetrahedron Lett. 2005, 46, 6605; (d) Northrup, A. B.; Mangion, I. K.; Hettche, F.; MacMillan, D. W. C. Angew. Chem., Int. Ed. **2004**, 43, 2152; (e) Córdova, A.; Ibrahem, I.; Casas, J.; Sundén, H.; Engqvist, M.; Reyes, E. Chem. Eur. J. 2005, 11, 4772; (f) Córdova, A.; Engqvist, M.; Ibrahem, I.; Casas, J.; Sundén, H. Chem. Commun. 2005, 2047; (g) Ibrahem, I.; Córdova, A. Tetrahedron Lett. 2005, 46, 3363; (h) Enders, D.; Grondal, C. Angew. Chem., Int. Ed. 2005, 44, 1210; (i) Suri, J. T.; Ramachary, D. B.; Barbas, C. F., III. Org. Lett. 2005, 7, 1383; (j) Westermann, B.; Neuhaus, C. Angew. Chem., Int. Ed. 2005, 44, 4077; (k) Enders, D.; Grondal, C.; Vrettou, M.; Raabe, G. Angew. Chem., Int. Ed. 2005, 44, 4079; (1) Córdova, A.; Notz, W.; Barbas, C.

F., III J. Org. Chem. 2002, 67, 301; (m) Northrup, A. B.; MacMillan, D. W. C. J. Am. Chem. Soc. 2002, 124, 6798; (n) Córdova, A. Tetrahedron Lett. 2004, 45, 3949.

- 9. (a) Sundén, H.; Ibrahem, I.; Eriksson, L.; Córdova, A. Angew. Chem., Int. Ed. 2005, 44, 4877; (b) Liao, W.-W.; Ibrahem, I.; Córdova, A. Chem. Commun. 2006, 674; (c) Sundén, H.; Ibrahem, I.; Córdova, A. Tetrahedron Lett. 2006, 47, 99; (d) Ibrahem, I.; Córdova, A. Angew. Chem., Int. Ed. 2006, 45, 1952; (e) Ibrahem, I.; Samec, J. S. M.; Bäckvall, J. -E.; Córdova, A. Tetrahedron Lett. 2005, 46, 3965.
- 10. For a recent review on one-pot asymmetric multi-component reactions see: Ramón, D. J.; Yus, M. Angew. Chem., Int. Ed. 2005, 44, 1602.
- 11. Compound 2a: $[\alpha]_D^{25} + 10.1$ (c 1.0, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$: δ 0.95 (t, $J = 7.6 \text{ Hz}, 3\text{ H}$), 1.09 (d, $J = 7.2$ Hz, 3H), 1.36–1.45 (m, 1H), 1.50–1.57 (m, 1H), 1.74 (br s, 1H), 2.39–2.45 (m, 1H), 3.43–3.48 (m, 1H), 3.72 (s, 3H), 5.86 (d, $J = 16.0$ Hz, 1H), 6.96 (dd, $J = 16.0$, 8.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 10.2, 16.0, 27.6, 42.4, 51.6, 76.3, 121.6, 151.2, 167.3. The ee was determined on acetylated 2a: Chiral GC: 93% ee, CP-Chirasil-DexCB column, temperature programme: 95 °C, hold 3 min, 95–120 °C, rate: 2 °C/min, hold 3 min, 120– 200 °C, rate: 80 °C/min, hold 5 min major isomer: $t_{\rm R}$ = 16.67 min; minor isomer: $t_{\rm R}$ = 17.81 min.
- 12. Kinoshita, K.; Takenaka, S.; Hayashi, M. J. Chem. Soc., Perkin. Trans. 1 1991, 2547.
- 13. To a stirred solution of (S)-proline (12 mg, 10 mol %) and isobutyraldehyde (144 mg, 2 mmol) in DMF (1 mL), propionaldehyde (58 mg, 1 mmol) was added slowly over 42 h at $+4$ °C. Next, the reaction mixture was allowed to reach room temperature and LiBr (217.5 mg, 2.5 mmol), methyl 2-(diethoxyphosphoryl)acetate (525 mg, 2.5 mmol) and DBU (380 mg, 2.5 mmol) were added. After 4 h of stirring at room temperature, the reaction mixture was passed through a silica gel FC column (ethyl acetate/ pentane $= 1:7$) to give the pure product 2b as a clear oil (108 mg, 58%). Compound **2b:** $\left[\alpha\right]_{25}^{25}$ 12.7 (c 1.0, CHCl₃);
¹H NMR (300 MHz, CDCl.); δ 0.94 (d, $I = 6.6$ Hz, 6H) ¹H NMR (300 MHz, CDCl₃): δ 0.94 (d, J = 6.6 Hz, 6H), 1.09 (d, $J = 6.9$ Hz, 3H), 1.67–1.75 (m, 1H), 2.50–2.58 (m, 1H), 3.19 (t, $J = 6.0$ Hz, 1H), 3.73 (s, 3H), 5.88 (dd, $J = 15.9, 0.9$ Hz, 1H), 7.01 (dd, $J = 15.9, 9.0$ Hz, 1H); ¹³C NMR (75 MHz, CDCl₃): δ 16.7, 17.0, 19.4, 30.9, 39.8, 51.4, 79.7, 121.4, 151.0, 167.0. The ee was determined on acetylated 2b: Chiral GC: 96% ee, CP-Chirasil-DexCB column, Temperature programme: 95 °C, hold 3 min, 95– 120 °C, rate: 1.2 °C/min, hold 3 min, 120–200 °C, rate: 80 °C/min, hold 5 min, major isomer: $t_R = 22.47$ min; minor isomer: $t_{\rm R} = 23.66$ min.

Compound 2c: $[\alpha]_D^{25}$ +13.3 (c 1.0, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 1.10 \text{ (d, } J = 6.8 \text{ Hz}, 3\text{H}), 1.04-1.87$ (br m, 11H), 2.56–2.60 (m, 1H), 3.20–3.23 (m, 1H), 3.73 (s, 3H), 5.87 (d, $J = 16.0$ Hz, 1H), 7.01 (dd, $J = 16.0$, 8.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 17.1, 26.1, 26.4, 26.6, 27.9, 29.8, 39.3, 41.0, 51.7, 79.4, 121.8, 151.1, 167.2. Chiral GC: 97% ee, CP-Chirasil-DexCB column, temperature programme: 110° C, hold 6.5 min, $110-160^{\circ}$ C, rate: 2 °C/min, hold 6.5 min, 160–200 °C, rate: 80 °C/min, hold 5 min, major isomer: $t_R = 35.259$ min; minor isomer: $t_{\rm R} = 34.954 \text{ min.}$

- 14. Compound 2d: $[\alpha]_D^{25} 5.1$ (c 1.0, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$: δ 7.38–7.26 (m, 10H), 6.98 (dd, $J = 16.0, 6.8$ Hz, 1H), 6.09 (d, $J = 15.6$ Hz, 1H), 4.61 (d, $J = 11.6$ Hz, 1H), 4.51 (s, 2H), 4.39 (d, $J = 12.0$ Hz, 1H), 4.08 (t, $J = 8.4$ Hz, 1H), 3.93–3.90 (m, 1H), 3.76 (s, 3H), 3.59–3.56 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 51.8, 70.5, 71.9, 72.4, 73.6, 78.7, 123.9, 127.97, 128.04, 128.1, 128.5, 128.60, 128.64, 137.7, 137.9, 145.1, 166.4; HPLC: 98% ee (Daicel Chiralpak AD, *i*-hexane/*i*-PrOH = 96:4, flow rate: 0.5 mL/min , $\lambda = 254 \text{ nm}$: major isomer: $t_{\rm R} = 81.06$ min; minor isomer: $t_{\rm R} = 72.00$ min.
- 15. (a) VanRheenen, V.; Kelly, R. C.; Cha, D. Y. Tetrahedron Lett. 1976, 17, 1973; (b) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483.
- 16. Compound 3d: $[\alpha]_D^{25} 9.2$ (c 0.6, CHCl₃); ¹H NMR (400 MHz, CDCl3): d 3.62–3.70 (m, 3H), 3.80 (s, 3H), 4.05–4.13 (m, 2H), 4.49–4.71 (m, 5H), 7.25–7.37 (m, 10H); ¹³C NMR (100 MHz, CDCl₃): δ 52.9, 70.8, 70.9, 73.0, 73.8, 73.9, 74.4, 78.0, 128.2, 128.2, 128.3, 128.7, 128.8, 137.7, 138.0, 174.2.
- 17. Compound $4d'$: $[\alpha]_D^{25} + 10.1$ (c 0.6, CHCl₃); literature: $[\alpha]_{D}^{24}$ +14.3 (c 1.01, CHCl₃) Takahashi, H.; Hitomi, Y.; Iwai, Y.; Ikegami, S. J. Am. Chem. Soc. 2000, 122, 2995. All the spectroscopic data were identical to Takahashi et al. ¹H NMR (300 MHz, CDCl₃): δ 3.64–3.70 (m, 2H), 3.99–4.05 (m, 1H), 4.12 (dd, $J = 6.9$, 3.6 Hz, 1H), 4.37– 4.41 (m, 2H), 4.51–4.59 (m, 4H), 4.63 (d, $J = 11.7$ Hz, 1H), 4.74 (d, $J = 11.7$ Hz, 1H), 4.86 (d, $J = 12.0$ Hz, 1H), 5.05 $(d, J = 12.0 \text{ Hz}, 1H), 7.17-7.32 \text{ (m, 20H)}.$
- 18. Compound 5: $[\alpha]_D^{25} 34.5$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 7.40–7.20 (m, 10H), 6.83 (dd, $J = 10.8$, 1.8 Hz, 1H), 5.96 (dd, $J = 10.8$, 1.2 Hz, 1H), 4.66–4.45 (m, 6H), 3.89 (dd, $J = 10.8$, 2.7 Hz, 1H), 3.73 (dd, $J = 10.8$, 2.1 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 162.8, 146.5, 137.8, 137.2, 128.8, 128.7, 128.5, 128.3, 128.1, 128.0, 120.8, 80.3, 73.8, 72.5, 69.0, 68.2. HPLC: 97% ee (Daicel Chiralpak AD, *i*-hexane/*i*-PrOH = 96:4, flow rate: 0.5 mL/min, $\lambda = 254$ nm): major isomer: $t_R = 40.28$ min; minor isomer: $t_R = 44.71$ min.
- 19. Compound 6: $[\alpha]_D^{25} 70.9$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 2.81 (br s, 1H), 3.34 (br s, 1H), 3.69–3.78 (m, 2H), 4.04–4.08 (m, 2H), 4.41–4.46 (m, 2H), 4.50–4.52 (m, 2H), 4.55–4.67 (m, 2H), 7.25–7.37 (m, 10H); ¹³C NMR (100 MHz, CDCl₃): δ 67.7, 68.1, 69.5, 71.9, 72.3, 73.8, 79.5, 128.0, 128.2, 128.4, 128.6, 128.7, 128.9, 137.1, 137.8, 172.8.
Compound $7 \cdot \lceil n^{25} \rceil$
- 20. Compound 7: $[\alpha]_D^2$ -1 (c 0.1, CHCl₃); literature, ent-7: $[\alpha]_{D}^{25}$ +0.37 (c 0.1, CHCl₃), Mahmud, T.; Tornus, I.; Egelkrout, E.; Wolf, E.; Uy, C.; Floss, H. G.; Lee, S. J. Am. Chem. Soc. 1999, 121, 6973. All the spectroscopic data were identical to Mahmud et al. ¹H NMR (300 MHz, CDCl₃): δ 3.67–3.76 (m, 3H), 3.89–3.92 (m, 1H), 3.97–4.00 (m, 1H), 4.18–4.21 (m, 1H), 4.38–4.42 (m, 2H), 4.51 (s, 2H), 4.61–4.71 (m, 2H), 4.83 (d, $J = 12.0$ Hz, 1H), 4.97 (d, $J = 12.0$ Hz, 1H), 7.17–7.32 (m, 20H).